## Advanced Statistical Physics Problem Set 7

 ${\rm due:~08.12.2010}$  available at http://userpage.fu-berlin.de/psilvest/

**Problem 7.1** Extension of problem 6.2 from the previous set: You probably already noticed that in this problem you were calculating the entropy of the ensemble of M harmonic oscillators

$$\mathcal{H}(\{n_i\}) = \sum_{i=1}^{M} \hbar \omega (n_i + \frac{1}{2}), \tag{1}$$

with the energy

$$E = \hbar\omega(N + \frac{M}{2}). \tag{2}$$

- (a) Continue the same calculation and find the probability p(n) that a particular oscillator is in its n-th level.
- (b) Find p(n) from the Canonical ensemble. Compare results for Canonical and Microcanonical ensembles.
- (c) Find the average energy  $\varepsilon$  per oscillator at a given temperature T. Use the Canonical ensemble and formulas from the lecture. [2p]

**Problem 7.2** The Canonical ensemble: Show that if the temperature is uniform, the pressure of a gas in a uniform magnetic field decreases with height according to  $P(z) = P(0) \exp(-mgz/k_BT)$ , where m is the mass of the gas molecule. [2p]

**Problem 7.3** Rubber molecule: A one-dimensional polymer molecule consists of N  $(N \gg 1)$  elements of length a. Let the distance between the ends of the molecule be x (see figure). Find the entropy of the molecule as a function of x and obtain the relation between the temperature of the chain and the force(tension) which is necessary to maintain the distance x at a temperature T, assuming the joints can turn freely.

Is not it amazing that a molecule with zero internal energy can cause a nontrivial tension?

Small hint: You may apply your knowledge of Statistical physics to find the Free energy and then use the standard machinery of Thermodynamics. [2p]

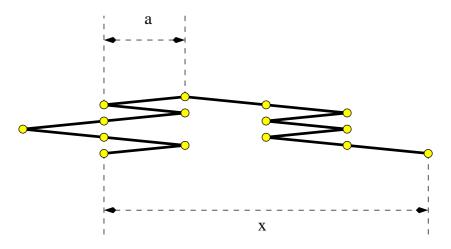


Figure 1:

**Problem 7.4** Molecular adsorption: N diatomic molecules are stuck on a metal surface of a square symmetry. Each molecule can either lie flat on the surface, in which case it has to be aligned to one of two directions, x and y, or it can stand up along the z direction. There is an energy cost of  $\varepsilon > 0$  associated with a molecule standing up and zero energy for molecules lying flat along x or y directions.

- (a) How many microstates have the smallest value of energy? What is the largest microstate energy?
- (b) What is the largest positive energy at any positive temperature? [2p] You (almost) do not need to do any calculation to answer the previous 2 questions. Now use the Microcanonical ensemble:
- (c) For microcanonical macrostates of energy E, calculate the number of states  $\Omega(E, N)$ , and the entropy S(E, N).
  - (d) Calculate the heat capacity C(T) and sketch it.
  - (e) What is the probability that a specific molecule is standing up? [2p]